# **Green Chemistry**

## **Question Paper**

Level	International A Level
Subject	Chemistry
Exam Board	Edexcel
Topic	Application of Core Principles of Chemistry
Sub Topic	Green Chemistry
Booklet	Question Paper

Time Allowed:

41 minutes

Score: /34

Percentage: /100

#### **Grade Boundaries:**

A*	Α	В	С	D	Е	U
>85%	'77.5%	70%	62.5%	57.5%	45%	<45%

**1** Glucose is fermented to produce ethanol.

	$C_6H_{12}O_6 \rightarrow 2C_2H_5OH + 2CO_2$									
What	What is the atom economy, by mass, for the production of ethanol in this reaction?									
(Relati	(Relative molecular masses: $C_6H_{12}O_6 = 180$ , $C_2H_5OH = 46$ , $CO_2 = 44$ )									
⊠ A	25.6%									
⊠B	48.9%									
<b>⊠</b> C	50.0%									
⊠ D	51.1%									
	(Total for Question 1 = 1 mark)									
<b>2</b> The fo	ollowing industrial processes produce ethanol.									
	Hydration of ethene: $C_2H_4(g) + H_2O(g) \xrightarrow{H_3PO_4} C_2H_5OH(I)$									
	Trydration of etherie: $C_2\Pi_4(g) + \Pi_2O(g)$ $\longrightarrow$ $C_2\Pi_5O\Pi(f)$									
	Fermentation of glucose: $C_6H_{12}O_6(aq)$ $\xrightarrow{yeast}$ $2C_2H_5OH(aq) + 2CO_2(g)$									
	hat is the atom economy by mass of the fermentation process?									
Us	se the Periodic Table as a source of data. (1)									
⊠ A	25.6%									
<b>В</b>	46.0%									
	51.1%									
⊠ D	92.0%									
	om a 'green chemistry' perspective, which of the following is an advantage of e hydration of ethene compared to the fermentation process?									
⊠ A	The catalyst is less corrosive.									
ВВ	A higher temperature is needed.									
	Ethene is a renewable resource.									
■ D	The ethanol is easier to purify.									
	(Total for Question 2 = 2 marks)									

**3** The relative greenhouse factor of a gas compares the greenhouse effect of a molecule of the gas to that of a molecule of carbon dioxide.

Gas	Percentage by volume in the atmosphere	Relative greenhouse factor per molecule
CO <sub>2</sub>	3.5 × 10 <sup>-2</sup>	1
CH <sub>4</sub>	1.7 × 10 <sup>-4</sup>	30
N <sub>2</sub> O	3.1 × 10 <sup>-5</sup>	200
CCI <sub>2</sub> F <sub>2</sub>	4.8 × 10 <sup>-6</sup>	25 000

Using **only** the data in the table above, select the gas that has the greatest greenhouse effect in the atmosphere.

X	Α	$CO_2$
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$$\square$$
 C N<sub>2</sub>O

$$\square$$
 **D**  $CCl_2F_2$ 

(Total for Question 3 = 1 mark)

4	Butane is an aerosol propellant now used as an alternative to CFCs. Although it is less destructive to the ozone layer, it has the disadvantage of being

**A** very corrosive.

**B** highly flammable.

**C** hard to evaporate.

**D** highly toxic.

(Total for Question 4 = 1 mark)

5	Which is the <b>best</b> ex	planation of why	carbon dioxide,	$CO_a$ , is a	greenhouse gas	;?

- A It is in high concentration and has a long residence time in the upper atmosphere so it absorbs infrared radiation significantly.
- **B** It is a polar molecule and so absorbs infrared radiation.
- ☑ C It absorbs ultra-violet radiation and re-emits infrared radiation.
- **D** It has polar bonds that absorb and re-emit infrared radiation.

(Total for Question 5 = 1 mark)

**6** The term 'carbon footprint' is concerned with the amount of carbon dioxide produced in generating a certain amount of energy.

The table below gives some data about several fuels.

Fuel	Energy density / MJ I <sup>-1</sup>	$CO_2$ produced on combustion / g $I^{-1}$
Petrol	32 2328	
Diesel	36	2614
LPG	24	1533
Bioethanol	21	1503

By calculating the mass of CO<sub>2</sub> produced per MJ of energy for each fuel, identify which fuel would give the **smallest** carbon footprint.

■ B Diesel

C LPG

**D** Bioethanol

(Total for Question 6 = 1 mark)

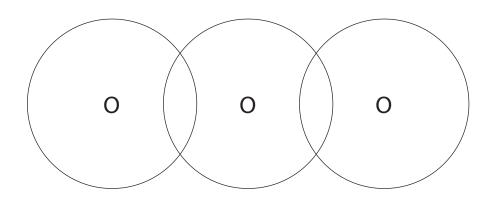
7	Nitrogen oxide, NO, can act as a catalyst in the depletion of the ozone layer. Which of the following reactions is most likely to be a step in the process?									
	×	A	2NO	+	20 <sub>3</sub>	$\rightarrow$	$N_2O_4$	+	40	
	×	В	2NO	+	20 <sub>3</sub>	$\rightarrow$	$N_2$	+	40 <sub>2</sub>	
	×	C	2NO	+	20 <sub>3</sub>	$\rightarrow$	2NO <sub>2</sub>	+	20 <sub>2</sub>	
	×	D	2NO	+	20 <sub>3</sub>	$\rightarrow$	$N_2O$	+	3½O <sub>2</sub>	
_										(Total for Question 7 = 1 mark)
8	Th	e gr	eenhou	ıse ga	as with	n the	largest a	vera	ge conce	ntration in the atmosphere is
	X	A	carbor	n dio	xide.					
	X	В	metha	ne.						
	☑ C nitrogen.									
	X	D	water	vapo	ur.					
										(Total for Question 8 = 1 mark)
9	Low molecular mass alkanes are now used as propellants in aerosols. Which environmental problem does this aim to reduce?									
	X	A	Acid ra	ain						
	X	В	Global	warı	ming					
	X	C	Non-b	iode	gradak	oility				
	X	D	Ozone	dep	letion					
										(Total for Question 9 = 1 mark)

10	Susta	inable chemistry aims to involve processes which use								
	⊠ A	non-renewable resources.								
	⊠ B	a catalyst.								
	⊠ C	high pressure.								
	⊠ D	high temperature.								
		(Total for Question 10 = 1 mark)								
11		e is serious concern over climate change brought about by anthropogenic . Which of the following is <b>not</b> one of these?								
	⊠ A	Burning of fossil fuels.								
	⊠ B	Deforestation.								
	⊠ C	☑ C Intensive agriculture.								
	☑ D Volcanic eruptions.									

- **12** This is a question about environmental chemistry.
  - (a) Ozone,  $O_3$ , is a non-linear molecule present in the Earth's upper atmosphere. It absorbs ultraviolet radiation from the Sun and so protects living organisms from this type of radiation.
    - (i) Complete the dot and cross diagram for the ozone molecule. Show the outer electrons only.

Use dots ( $\bullet$ ) for the electrons of the left-hand oxygen atom, crosses ( $\mathbf{x}$ ) for the central oxygen atom and triangles ( $\Delta$ ) for the right-hand oxygen atom.





(ii) Explain why ozone is a non-linear molecule.	(1)
 (iii) State <b>one</b> harmful consequence to a person of increased exposure to ultraviolet radiation.	(1)
 (iv) What property of ultraviolet radiation makes it more harmful than infrared radiation to living organisms? Justify your answer.	(1)

<ul><li>(v) Nitrogen oxides in aircraft emissions are involved in the depletion of the ozone layer. One of these oxides is the free radical nitrogen monoxide.</li></ul>		
Define the term <b>free radical</b> .	(1)	
(vi) Complete the equations below for the reaction of the nitrog with ozone.	en monoxide (3)	
Reaction 1 NO $^{\bullet}$ + O <sub>3</sub> $\rightarrow$ +		
Reaction 2 $\qquad \qquad + O_3 \rightarrow \qquad + \qquad \qquad + \qquad \qquad$		
Overall Reaction $\rightarrow$		
(vii) Based on your answer in (a)(vi), what is the role of the nitrog the depletion of the ozone layer?	en monoxide in (1)	
(viii) Suggest why the release of free radical nitrogen oxides by ve buses and lorries, does not affect the ozone layer.	ehicles, such as	
(b) Environmentalists are concerned by the increase in concentration atmosphere of gases that absorb infrared radiation because of the climate change.		
(i) Carbon dioxide is a molecule in the atmosphere that absorbe radiation.	s infrared	
Explain why this molecule absorbs infrared radiation and what absorption has on the molecule.	eat effect this (2)	

(ii) Suggest a gas, of which there is a significant concentration in the atmosphere, which does <b>not</b> absorb infrared radiation.	(1)
(iii) CFCs make a significant contribution to global warming, despite being present in only very small concentrations in the atmosphere. Suggest a reason for this.	
(iv) Suggest why there is now little concern over the contribution of CFCs to global warming compared with that of carbon dioxide.	(1)
(v) Water vapour is another molecule in the atmosphere that absorbs infrared radi but it is not considered to be responsible for anthropogenic climate change. Juthis statement.	
(vi) The term 'carbon neutrality' has become widely used with reference to biofuels. Use of biofuels is one of the measures employed in an attempt to stabilise the level of carbon dioxide in the atmosphere and hence to reduce	
climate change.  Explain the term 'carbon neutrality' and suggest why biofuels are unlikely to be completely carbon neutral.	(2)

(vii) Suggest **two** measures, other than the use of biofuels, by which the chemical industry could reduce its carbon footprint.

2		(Total for Question 12 = 22 marks)
	2	
1		
(2)	1	
		(2)