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## **Experimental Techniques** Mark Scheme 1

| Level      | IGCSE                    |
|------------|--------------------------|
| Subject    | Chemistry                |
| Exam Board | CIE                      |
| Торіс      | Experimental Techniques  |
| Sub-Topic  |                          |
| Paper Type | Alternative to Practical |
| Booklet    | Mark Scheme 1            |

| Time Allowed: | 54 minutes |
|---------------|------------|
| Score:        | /45        |
| Percentage:   | /100       |

| Question     | Answer   | Marks | Guidance                                     |
|--------------|--|-------|--|
| 1 <b>(a)</b> | b line/origin clearly labelled on diagram;                                       | 1     |  |
| 1(b)         | any organic solvent/ethanol/alcohol/acetone;                                     | 1     | water/acids                                  |
| 1 <b>(c)</b> |  | 1     |  |
| 1(d)         | 1 and 3 present;<br>2 not present;<br>unknown dye present;                       | 3     | I reference to properties of dyes 1, 2 and 3 |
| 1(e)         | rep the experiment/use a different solvent/measure <i>R</i> <sub>f</sub> values; | 1     |  |

| 2 | (a  | a thermometer (1)   |     |  |
|---|-----|---|-----|--|
|   |     | condenser (1)<br><b>allow</b> condensing tube, condensating tube, etc.        | [2] |  |
|   | (b) | arrows labelled – water (in) and water (out) (1)                              | [1] |  |
|   | (c) | fractional (1)  |     |  |
|   |     | distillation (1)  | [2] |  |
|   | (d) | ethanol (1)   | [1] |  |
|   |     | (ii) temperature would rise (above 78°C) (1)                                  | [1] |  |
|   | (e) | alcohols are (in)flammable / catch fire / burn (1)<br><b>ignore</b> : explode |     |  |
|   |     | Bunsen burner / flame / heat (1)  | [2] |  |
|   |     |   |     |  |
| 3 | (a  | precipitation / double decomposition (1)<br><b>allow:</b> ppt                 | [1] |  |
|   | (b) | low / insoluble / does not dissolve (1)                                       | [1] |  |
|   |     | (ii) high / soluble / dissolves (1)   | [1] |  |
|   | (c) | filtration (1)  | [1] |  |

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| (a  | boxes completed to show stirrer / glass rod (1)<br>watchglass / evaporating dish (1)                                     | [2] |
|-----|--|-----|
| (b) | to speed up the reaction (1)   | [1] |
| (c) | correct answer 4.2g (2)<br>if incorrect, evidence of 17.8 – 13.6 (1)   | [2] |
| (d) | <ul> <li>(i) solid / lead oxide visible / remaining (1)</li> <li>do not allow: mention of precipitate</li> </ul>         | [1] |
|     | (ii) filtration (1)  | [1] |
|     | (iii) excess (1)<br>allow: residue   | [1] |
| (e) | Any <b>two</b> from:<br>evaporation / steam (1)<br>solid / crystals formed (1)<br>breakdown / decomposition of solid (1) | [2] |

4

max [6]

| 5 | (a)        | to p | prevent air / oxygen / bacteria entering jar (1)   | 11 <sup>J</sup> |
|---|------------|------|--|-----------------|
|   | (b)        | pes  | tle and / or mortar (1)  | 11 <sup>J</sup> |
|   | (c)        | dia  | gram of funnel and filter paper (1) labelled (1)   | [2]             |
|   | (d)        |      | st would not work at high temperatures/ kills yeast / denatures enzymes/ owtte (1)<br>ow: kills enzyme | 11 <sup>J</sup> |
|   | <b>(e)</b> | (i)  | bubbles/ froth (1)<br>not: gas/ CO2 given off/ turns cloudy  | 11 <sup>J</sup> |
|   |            | (ii) | (collect gas) and measure volume/ count bubbles (1)  |                 |
|   |            |      | over certain time interval (1)<br>allow: one mark for timing until bubbles/ reaction stopped           | [2]             |
|   | (f)        | frac | tional distillation (1)  | 11 <sup>J</sup> |

6 mass of silica gel (1) heat in oven > 100 °C (1) for specified realistic time / until turns blue (1) reweigh (1) record (1) heat in oven again to check constant mass / indication of colour change (1) calculation (1)