Thermal Properties and Temperature

Mark Scheme 4

Level	IGCSE
Subject	Physics
ExamBoard	CIE
Topic	Thermal Physics
Sub-Topic	Thermal Properties and Temperature
Paper Type	(Extended) Theory Paper
Booklet	Mark Scheme 4

Time Allowed: 49 minutes

Score: /41

Percentage: /100

1	(a	linea	sitive to box 5 ar to box 3 e range to box 2	B1 B1 B1	[3]	
	(b)	2 different metals (need not be named but must be identified as different) volt/millivolt/am/milliammeter/galvanometer/display reading V/mV/A/mA/°C AND circuit would work do not allow unlabelled box/meter ignore hot/cold junction labels		M1		
				A1	[2]	
			Ignore ca	s will not melt/gives p.d. at high temperature/remote sensing an withstand/will not be damaged by high temperature neat capacity/mass	B1 B1	[2]
2	(a)	less	s heat tra	not so hot (to hold) nsfer/sensible comment about air gap/more or better insulation explanation involving vacuum	B1 B1	[2]
	(b)			80) always above original line and below 80°C, reaches 5 min ends, straight or concave up, reaches 10 min	M1 A1	[2]
	(c)	red red exp exp	uces/stop uces/stop lanation lanation	om: os (energy losses by) convection os (energy losses by) evaporation os (energy losses by) radiation of mechanism of heat loss (by convection, evaporation or radiation) plus something like "which reduces heat losses" scores 2/2 on this of do more than restate question	B1 B1	[2]

(a	ΔT	= 50		C1 C1 A1	[3]
(b)			× 60)	C1 A1	[2]
(c)	acc ign	cept power for energy but not wrong/mixed quantities. Accept useful for total for input	or output,	C1 A1	[2]
(d)		·		B1	[1]
(e)	doe hig	esn't work at night/cloud cover/no sun/variable output h (initial) cost (of panels)		B1	[1]
(a) F	(l=)	E/m OR E/0.36 OR Pt/m OR Pt/0.36		C1 C1 A1	
(b)		liquid ignore vapour/gas/water		A1	
	(ii)	move around more rapidly / faster / more KE ignore start to vibrate etc but accept starts to vibrate faster move further apart / spreads out (NOT molecules expand) break free / evaporate / overcome bonds / overcome forces of attraction /escape / change state (accept boils) convection (current)	any 2	B1	[6]
	(b) (c) (d) (e)	(b) use end (c) effi accign effi (d) sou Giv (e) one doe hig do (a) Pt (l=) 3 ×	 ΔT = 50 Q = 798 000 J (b) use of E = Pt OR 170 × 8 OR see 1 360 OR energy = (170 × 8 × 3 600) = 4 896 000 J (c) efficiency = output(energy)/input (energy) OR his (a) + his (b) accept power for energy but not wrong/mixed quantities. Accept useful frignore total for input efficiency = 0.16 or 16% ecf from 6(a) and 6(b) (d) source not finite/will not run out ignore can be re-used/replaced Give for right idea e.g. accept sun always shines (e) one point from: doesn't work at night/cloud cover/no sun/variable output high (initial) cost (of panels) do not accept too low unless appropriate for a clearly stated context (a) Pt OR 1.2 × 10⁴ × 9 OR 1.2 × 10⁴ × (11 – 2) (I=) E/m OR E/0.36 OR Pt/m OR Pt/0.36 3 × 10⁵ J/kg (b) liquid ignore vapour/gas/water (ii) move around more rapidly / faster / more KE ignore start to vibrate etc but accept starts to vibrate faster move further apart / spreads out (NOT molecules expand) break free / evaporate / overcome bonds / overcome forces of attraction /escape / change state (accept boils) 	 ΔT = 50 Q = 798 000 J (b) use of E = Pt OR 170 × 8 OR see 1 360 OR × 60) energy = (170 × 8 × 3 600) = 4 896 000 J (c) efficiency = output(energy)/input (energy) OR his (a) ÷ his (b) accept power for energy but not wrong/mixed quantities. Accept useful for output, ignore total for input efficiency = 0.16 or 16% ecf from 6(a) and 6(b) (d) source not finite/will not run out ignore can be re-used/replaced Give for right idea e.g. accept sun always shines (e) one point from: doesn't work at night/cloud cover/no sun/variable output high (initial) cost (of panels) do not accept too low unless appropriate for a clearly stated context (a) Pt OR 1.2 × 10⁴ × 9 OR 1.2 × 10⁴ × (11 – 2) (i=) E/m OR E/0.36 OR Pt/m OR Pt/0.36 3 × 10⁵ J/kg (b) liquid ignore vapour/gas/water (ii) move around more rapidly / faster / more KE ignore start to vibrate etc but accept starts to vibrate faster move further apart / spreads out (NOT molecules expand) break free / evaporate / overcome bonds / overcome forces of attraction / escape / change state (accept boils) 	AT = 50 Q = 798 000 J A1 (b) use of E = Pt OR 170 × 8 OR see 1 360 OR × 60) C1 energy = (170 × 8 × 3 600) = 4 896 000 J A1 (c) efficiency = output(energy)/input (energy) OR his (a) + his (b) accept power for energy but not wrong/mixed quantities. Accept useful for output, ignore total for input efficiency = 0.16 or 16% ecf from 6(a) and 6(b) (d) source not finite/will not run out ignore can be re-used/replaced Give for right idea e.g. accept sun always shines B1 (e) one point from: doesn't work at night/cloud cover/no sun/variable output high (initial) cost (of panels) do not accept too low unless appropriate for a clearly stated context B1 (a) Pt OR 1.2 × 10 ⁴ × 9 OR 1.2 × 10 ⁴ × (11 – 2) (i=) Elm OR El0.36 OR Pt/m OR Pt/0.36 C1 3 × 10 ⁵ J/kg A1 (b) liquid ignore vapour/gas/water A1 (ii) move around more rapidly / faster / more KE ignore start to vibrate etc but accept starts to vibrate faster move further apart / spreads out (NOT molecules expand) erap attraction / escape / change state (accept boils)

5	(a	330 J of heat / energy required to change 1 g of ice to water at constant temperature / at melting point / at 0 degrees C			
	(b)	(i)	(B to C ice is) changing to water / melting / changing to liquid / changing state	B1	
			(D to E water is) changing to steam / vaporising / boiling / changing to gas	В1	
		(ii)	Sp. latent of vaporisation of water is greater than sp. latent of fusion of ice	B1	
		(iii)	s.h.c. of ice is less than s.h.c. of water	В1	
			more heat required to raise temperature of water OR rate of temperature rise of water is slower OR temperature rise of water takes longer	B1	[6]
6	(a	(i)	most: gas least: solid both required	B1	
		(ii)	because change of pressure (also) causes volume change (in a gas) NOT 'gas can be compressed'	B1	
	(b)		two from: expands uniformly (over required range) remains liquid over required range expands more than glass / has high expansivity / expansion has (reasonably) low specific heat capacity. has low freezing point / lower freezing point than mercury	x B2	
		(ii)	make (capillary) tube narrower (and longer) / thinner / smaller diameter make bulb larger (and tube longer) allow 'bore' for tube ignore 'smaller' ignore narrow thermometer	B1 B1	
	(c)	OR OR fast OR	t response) \times \text{\text{heat transfer more efficient / faster}}	D.	r
			glass takes up less heat ore reference to sensitivity ignore 'easier'	B1	[7]