

Density and Pressure

Mark Scheme 4

Level	IGCSE(9-1)
Subject	Physics
Exam Board	Edexcel IGCSE
Module	Double Award (Paper 1P)
Topic	Solids, Liquids and Gases
Sub-Topic	Density and Pressure
Booklet	Mark Scheme 4

Time Allowed: 65 minutes

Score: /54

Percentage: /100

Grade Boundaries:

A*	A	B	C	D	E	U
>85%	775%	70%	60%	55%	50%	<50%

Question number	Answer	Notes	Marks
1 (a) (i)	pressure = $\frac{\text{force}}{\text{area}}$	Allow symbols and rearrangements e.g. $p=F/A$	1
(ii)	substitute; rearrange; evaluate; matching unit; e.g. $270\,000 = F \div 0.016$ 1 mark $F = 270\,000 \times 0.016$ 2 marks 4320 3 marks N 4 th mark	Substitution and rearrangement in either order allow in words Allow alternatives with matching unit, e.g. 4.32 3 marks kN 4 th mark	4
(b)	Any three of MP1. idea of (continuous) random movement; MP2. collisions / impact/eq; MP3. with (inside) walls (of tyre); MP4. idea that force is produced (by bombarding molecules); MP5. idea of pressure as force on an area;	Allow momentum or NIII argument	3

(c)	<p>any three of-</p> <p>MP1. (now) more particles/molecules in the tyre;</p> <p>MP2. molecules have more speed /more energy (because gas is warmer);</p> <p>MP3. more impacts/more frequent impacts / harder impacts (with walls of tyre);</p> <p>MP4. (hence) more force on the inside;</p>	<p>Allow change of momentum argument</p> <p>Allow collisions with walls</p> <p>do not award MP3 if the impacts are only with other molecules</p>	3
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Total 11 marks

Question number	Answer	Notes	Marks
2 (a) (i)	<p>density = $\frac{\text{mass}}{\text{volume}}$</p>	<p>Allow symbols and rearrangements, e.g. $\rho = m / V$</p>	1
(ii)	<p>substitution into correct equation; calculation; matching unit; e.g. Density = $138 \div 16.3$ = 8.47 g/cm³</p>	<p>8.466, 8.5</p>	3
(b)	<p>B (incorrect and slightly too small)</p>		1

Total 5 marks

Question number			Answer	Notes	Marks
3	(a)	(i)	substitution / rearrangement; final value for volume; final value for time; e. $8 \times 200 = V \times 1$ $V = 1600$ (litres) time = 100 (minutes)	$(p_1V_1 = p_2V_2)$ – no mark as given on page 2. No credit for merely quoting the equation. Allow 99 minutes (i.e. assumption that the final 16 litres not available)	3
		(ii)	Any two suitable points, e.g. MP1. pressure decreases as depth decreases; MP2. reference to $p = h\rho g$; MP3. reference to pV equation (if temperature constant); MP4. additional bubbles join together as they rise; MP5. temperature increases nearer surface;		2
3	(b)	(i)	displacement method described; measure water displaced (with measuring cylinder); OR measure radius / diameter / circumference; calculate volume (with equation);		2
		(ii)	not a fair test; change of temperature / volume;	ignore 'each pump will have different pressure'	2

Total 9 marks

Question Number	Answer		Marks
4 (c) (i)	density = mass / volume;	ALLOW standard symbols (ALLOW d for density) VALUE MUST be 2 s.f. to be given evaluation mark 2300 if unit is kg/m ³	1
(ii)	substitution into correct equation: evaluation: unit: e. 54/23 2.3 g/cm ³		1 1 1
(d) (i)	compare with / look it up in; a book / data table / internet;		1 1
(ii)	any ONE suitable, e.g. (many) rock types with similar / same values; uncertainty in value / inaccurate measurements; data tables incomplete;		1 IGNORE human error ALLOW 'rock may not be pure'
		Total	12

Question number	Answer	Notes	Marks
5 (a)	ANY THREE of particles in constant motion / particles have kinetic energy; in random directions; colliding with walls; causing a force on the walls; Pressure = force /area;	Answers need to refer to particles / molecules rather than 'the gas is...' ALLOW 'Hitting the walls' / 'bouncing off the walls' ALLOW 'push' / 'pushing'	3
(b) (i)	(pressure would) increase;		1
(b) (ii)	(higher temp) increases (average) speed / kinetic energy of particles; So collide with walls more often / at higher speed;	IGNORE references to 'heating the particles' ALLOW 'hit harder'	1 1
(c)	Use of $p_1V_1 = p_2V_2$ (equation given) /substitution; 2000 (cm ³);	2000 alone scores 2	2
		Total	8

Question number	Answer	Notes	Marks					
6 (a)	density = mass/volume	ACCEPT equivalent rearrangement ACCEPT suitable abbreviations e.g. $\rho = m/v$ or $d = m/v$ REJECT equation 'triangles' alone	1					
(b)	D		1					
(c)	<table border="1" data-bbox="310 638 1062 922"> <thead> <tr> <th data-bbox="310 638 695 732">Measuring instrument</th> <th data-bbox="695 638 1062 732">Quantity measured</th> </tr> </thead> <tbody> <tr> <td data-bbox="310 732 695 826">measuring cylinder</td> <td data-bbox="695 732 1062 826">volume</td> </tr> <tr> <td data-bbox="310 826 695 922">electronic balance</td> <td data-bbox="695 826 1062 922">mass</td> </tr> </tbody> </table>		Measuring instrument	Quantity measured	measuring cylinder	volume	electronic balance	mass
Measuring instrument	Quantity measured							
measuring cylinder	volume							
electronic balance	mass							

Question number	Answer	Notes	Marks
6 (d)	<p>MAX TWO FOR EACH</p> <p>measuring cylinder – eyes to water level / perpendicular view; to avoid parallax; measurement at bottom of meniscus; measuring cylinder on flat surface / clean cylinder;</p> <p>electronic balance – place on stable surface /avoid disturbing balance; set to zero / check zero; finding mass without an with water – (tare or subtraction);</p>	<p>Ignore repetition wherever seen</p> <p>Ignore clean balance</p>	4
(e) (i)	<p>temperature / type of water (e.g. salinity, not 'heavy')</p>	<p>DO NOT ACCEPT answers referring to keeping the apparatus the same</p>	1
(ii)	<p>can also affect the density / volume (DOP)</p>	<p>ACCEPT arguments that follow through e.g. increasing temperature will increase the volume, therefore decreasing the density REJECT idea that mass is affected by change in temperature</p>	1